

ASBC Annual Meeting

June 4–7 ■ Fort Myers, Florida

See what SCIENCE can brew for you

Bitterness, Perception of Taste and Aroma

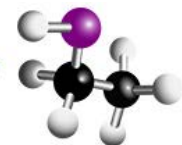
Pattie Aron

Rahr Malting

June 2, 2017

Bitterness

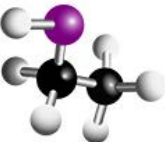
- Bitterness Perception
- Factors that affect perception
- Sources – Main components
- Sensory evaluation and techniques



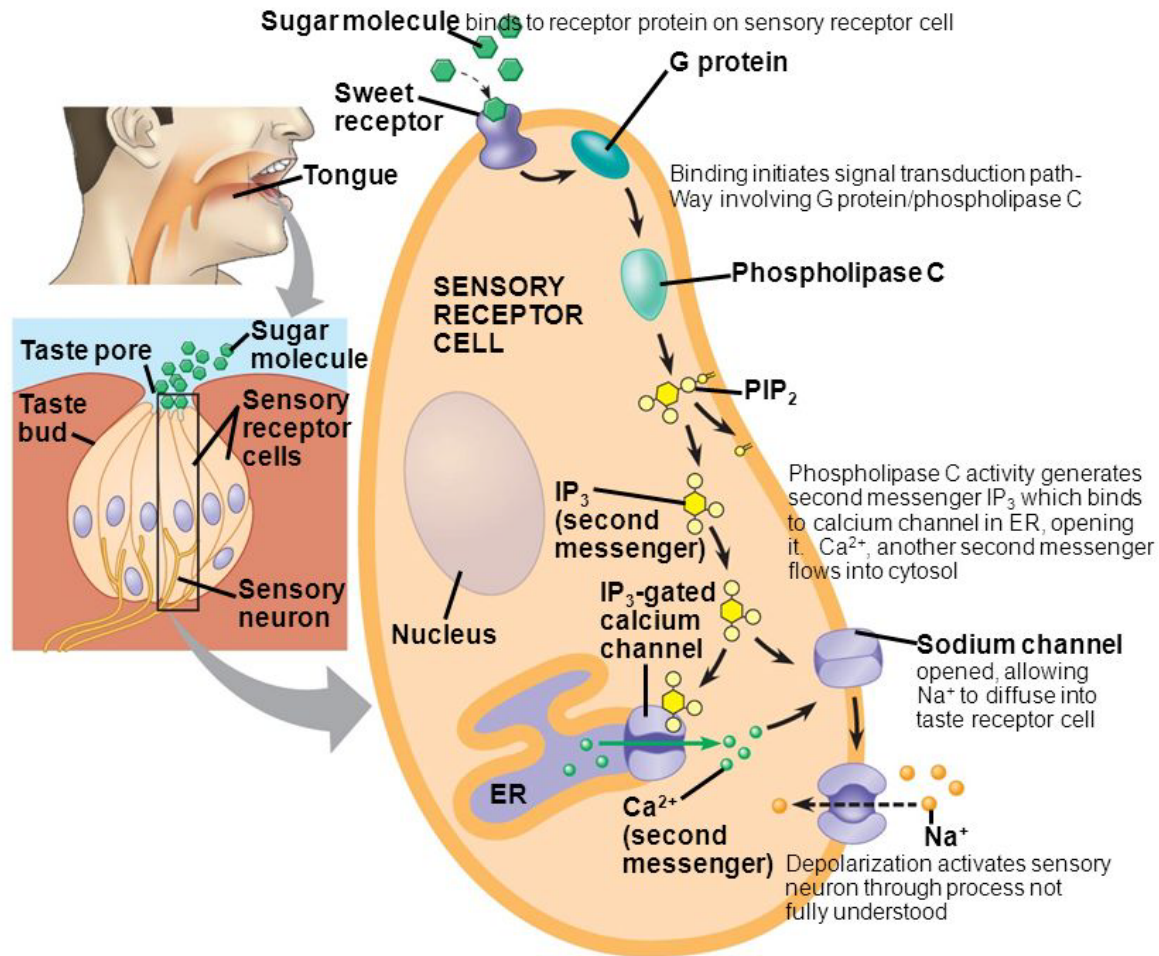


Bitterness Perception

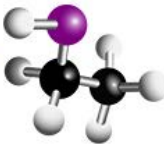
- Molecules bind to receptors on the tongue
 - Type II receptor cells (sweet, umami, bitter)
- Ligand binding site changes shape
- Interacts with a G-protein coupled receptor (GPCRs)
- G- protein activates messenger cell...cascade effect.....
- Ion channels activate and cell gradient changes
- Nerve cell stimulated
- Signals the Brain
- ***BITTER!***



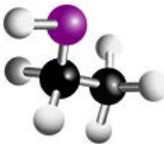
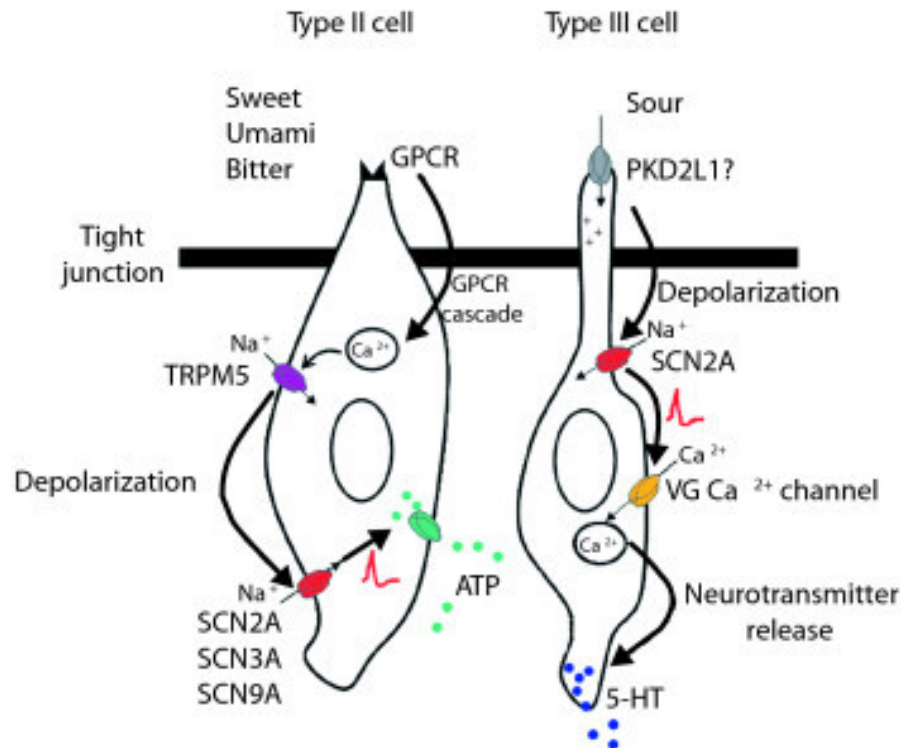
G – Protein Response



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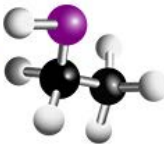
Type II Receptor





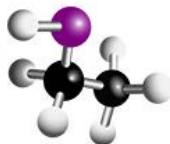
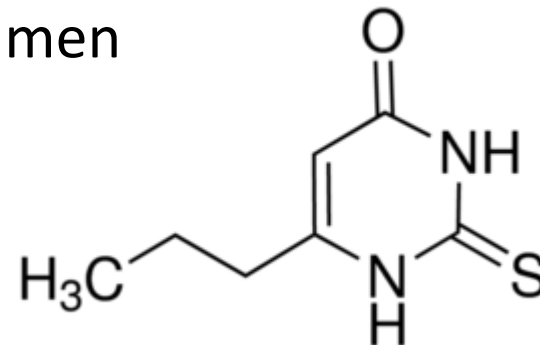
Bitterness – Perception Factors

- **Genetics –**
 - Heritage – 25 Taste Type II Receptors
 - Cluster of genes located on c 5p, 7q, 12 p
 - # of fungiform papillae vary by individual
 - Sex
- **Age** - response declines with age
- **Diet** - brain response change due to ‘training’
- Presence of **suppressants and enhancers**
 - Sugar
 - Salt
 - Acids
 - Metals



Bitterness Genetics and PROP

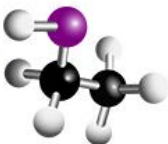
- PROP (6-propyl-2-thiouracil)
 - Bitter receptor – TAS2R38
 - Linked to chromosome locus at 5p15
 - Dominant trait
 - 70% of Caucasians are sensitive
 - 90% of Asians and African Americans
 - Subgroup – supertasters
 - More women than men



Bitterness – Beer Contributors

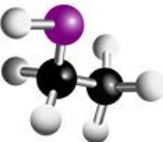
- **Hop Oils** Oil ‘burn’ may enhance bitterness
- **Polyphenols** - bitter
- **Color** (malt roast) – bitter compounds
- **Alcohol** may enhance or reduce
- **Higher pH** - enhances bitterness
- **Mineral Content**
 - Burtonization – sulfate = crisp
 - Carbonate – broader, harsher bitter
- **Carbonation bite** can enhance bitterness
- **Aging** decreases bitterness

- **Hop Acids and their products...**

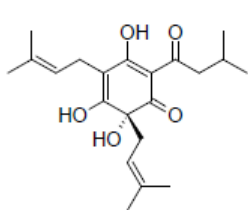


Bitterness - Hop Acids

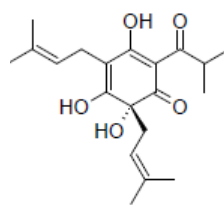
- Alpha Acids
- Beta Acids
- Isomerized alpha acids
- Reduced hop acids



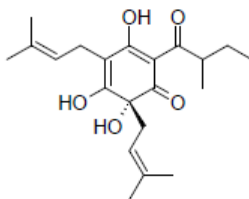
Alpha Acid Analogues



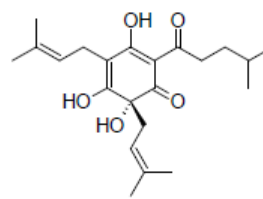
Humulone



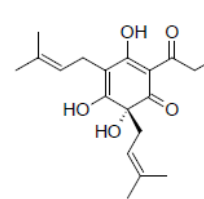
Cohumulone



Adhumulone



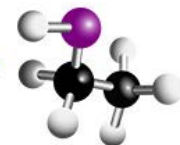
Prehumulone



Posthumulone

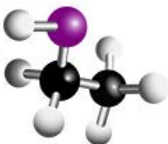
Alpha Acid	Acyl R	MW	%
Humulone*	$\text{CH}_2\text{CH}(\text{CH}_3)_2$	362	35-70
Cohumulone*	$\text{CH}(\text{CH}_3)_2$	348	20-65
Adhumulone**	$\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$	362	10-15
Prehumulone***	$\text{CH}_2\text{CH}_2\text{CH}(\text{CH}_3)_2$	376	1-10
Posthumulone***	CH_2CH_3	334	1-3

(Rigby, Bethune, *1952, **1953 and ***Verzele 1955)



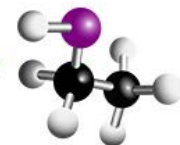
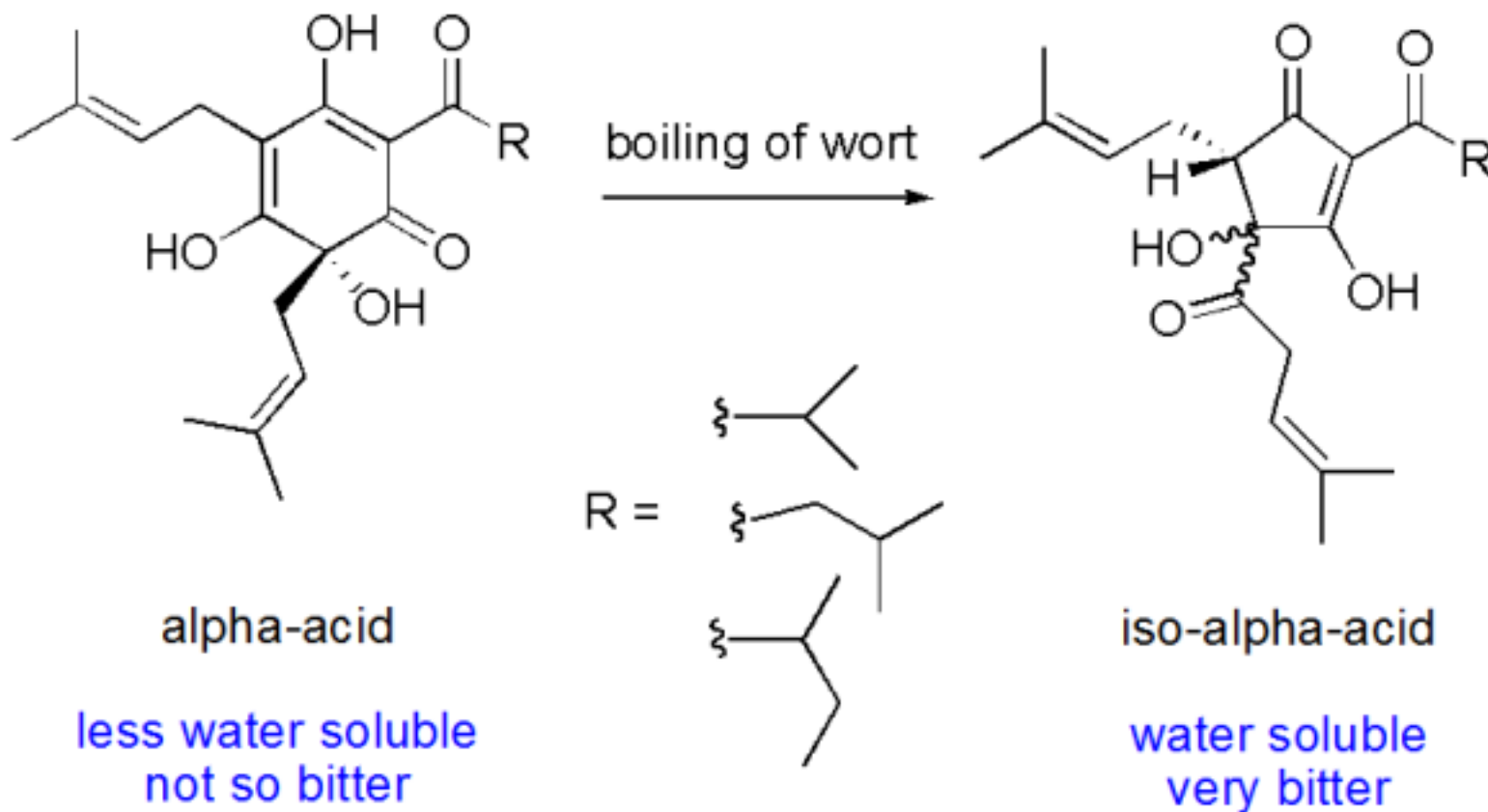
Bitterness - Alpha Acids

- Not bitter.
- Unstable – oxidize readily in presence of oxygen, heat and light.
- Some oxidized alpha acids form hard resins that do not contribute to beer bitterness.
- Some oxidized alpha acids do contribute to bitterness: humulinones and humulinic acids.
- At 25°C humulone aqueous solubility is low ~6mg/L
- Alpha acids are relatively insoluble in wort at pH 5, reaching a maximum of about 84 ppm when heated at pH 5.2, and even higher at pH ~6.5.



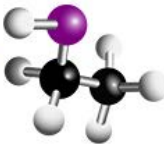
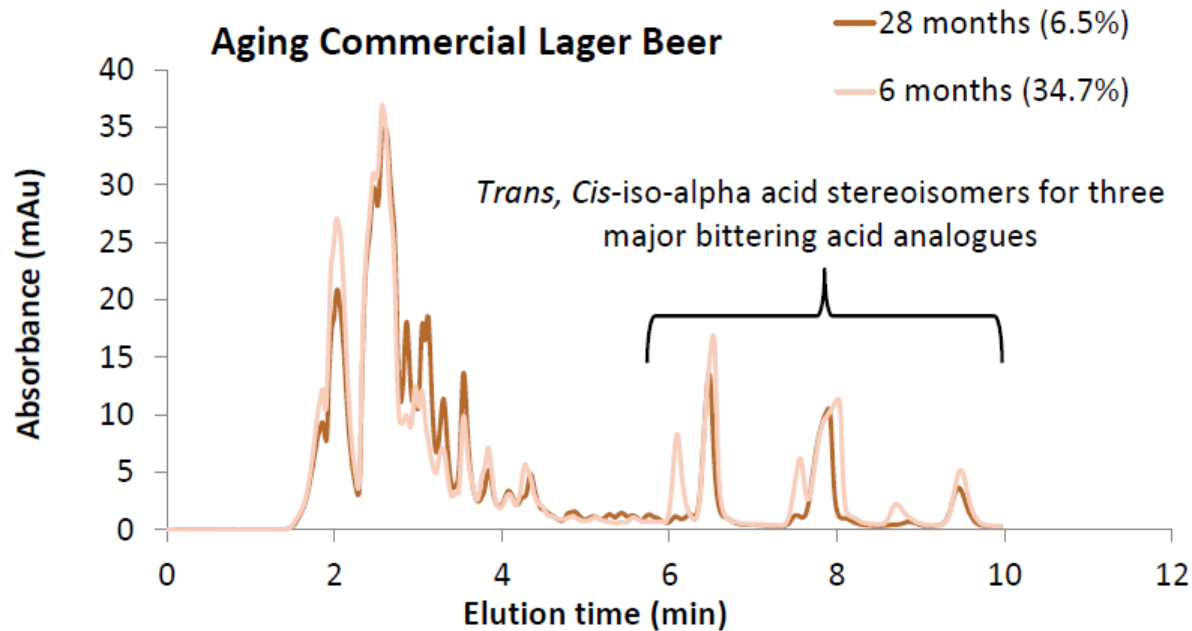
Isomerized alpha acids: Iso humulones

Isomerization



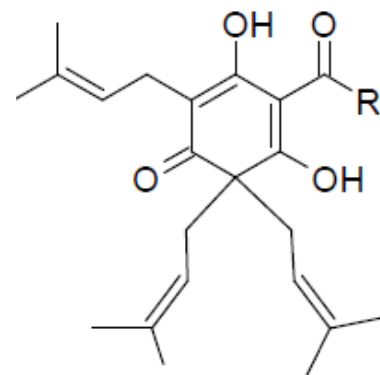
Iso-alpha acid stability

Thermal instability - Cis is thermally more stable, losses of trans occur over time

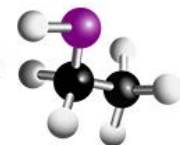


The Beta Acid Analogues

Beta Acid	Acyl R	%
Lupulone*	$\text{CH}_2\text{CH}(\text{CH}_3)_2$	30-55
Colupulone*	$\text{CH}(\text{CH}_3)_2$	20-55
Adlupulone**	$\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$	10-15
Prelupulone***	$\text{CH}_2\text{CH}_2\text{CH}(\text{CH}_3)_2$	1-3
Postlupulone***	CH_2CH_3	?

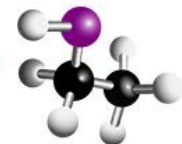


Beta Acids



Bitterness - The Beta Acids

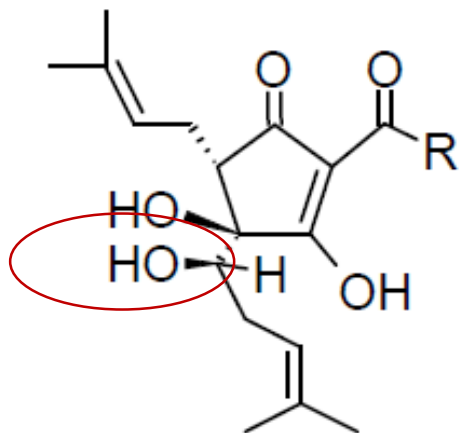
- Not bitter unless oxidized.
- Poorly soluble in water and wort.
- Poor solubility as pure compounds (1 g/100mL), but more soluble as a mixture
- Stable to alkaline hydrolysis in absence of oxygen.
- Susceptible to oxidation comparable to alpha acids
- Oxidation results in hulupones – products have ‘undesirable?’ bitterness and can make up for loss of alpha in old hops.



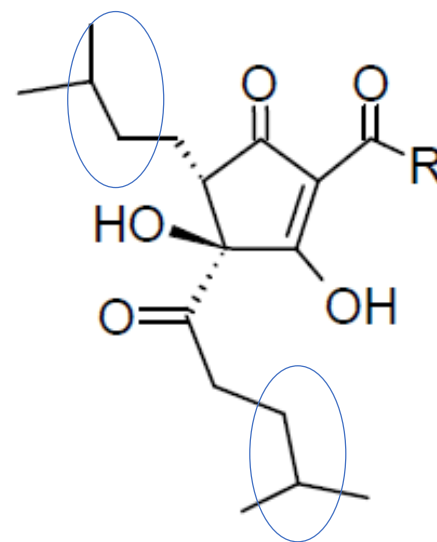
Advanced Hop Acids

Reduced Iso-alpha acids (Rho)

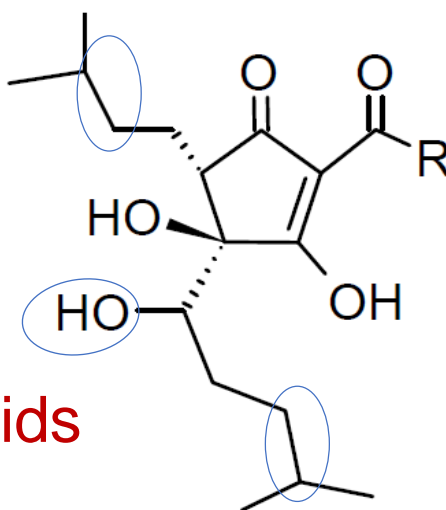
Trans-DIHYDRO-ISOHUMULONES or *Trans*-RHO-ISOHUMULONES



Tetrahydro-Iso-alpha acids



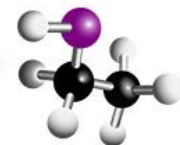
Hexahydro-Iso-alpha acids



Trans-tetrahydro-isocohumulones

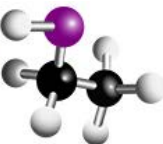
Hexa-hydro-isocohumulones

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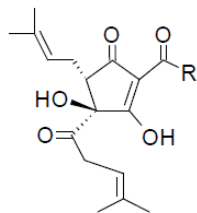
Bitterness – Sensory Evaluation

- Bitterness and the IBU
- Bitterness and Quality
- Bitterness Intensity
- Lingering – Time Intensity

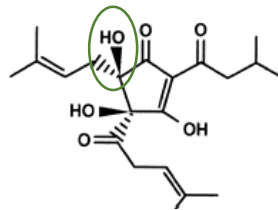


Sensory and the IBU

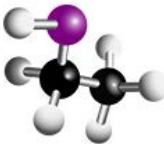
- The IBU measurement includes:
 - Iso-alpha acids, α , β , and oxidized products:
 - Humulinones – more soluble than IAA
 - Beta acid derivatives
 - Other hard resin derivatives
 - Anything soluble in isooctane that also absorbs near 275 nm under acidified conditions:
 - Phenolics: xanthohumol, flavonoids, etc.



Iso- alpha acid



OXI- Iso- alpha acid
(humulinone)

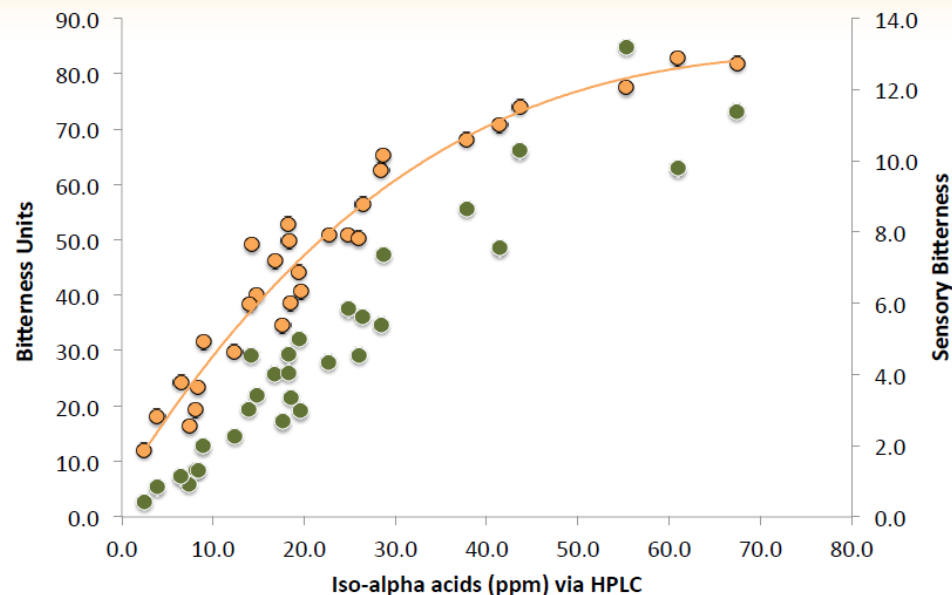


Contributions to the IBU: Sensory

Sensory bitterness will vary based on hopping rates, hopping technology, age of hops, variety of hops.

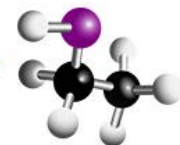
Qualitative differences may not correlate to IBU

Sensory bitterness does not track BU/IAA

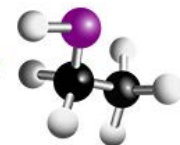
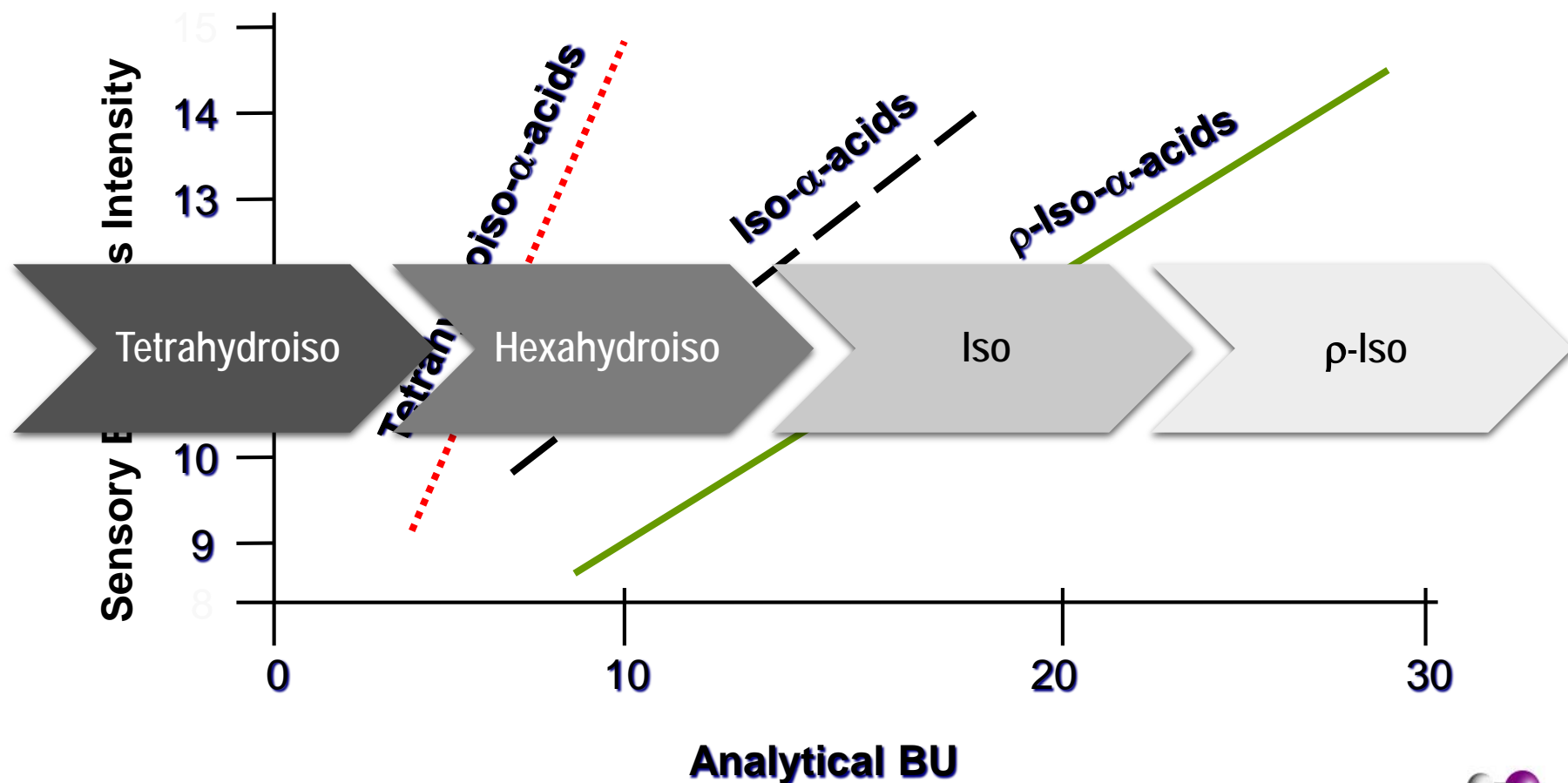


Source: WBC 2014, T. H. Shellhammer, Dry hopping contributions to bitterness

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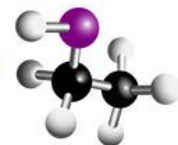


Comparison of Sensory vs. Analytical Bitterness



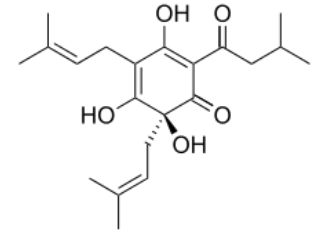
Bitterness Perception - Quality

- Harmonious
- Harsh
- Vegetative
- Medicinal
- Short
- Lingering

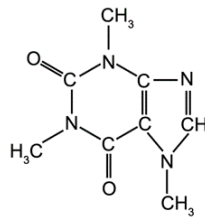


Bitterness Quality – Training Reference Types

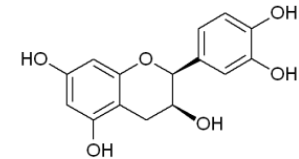
- Iso - alpha acids
 - Beer ranges from 2 ppm to 40 ppm
 - Higher ppm may occur with increased alcohol in specialty beers



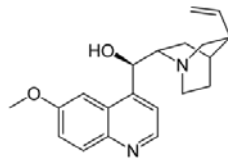
- Caffeine



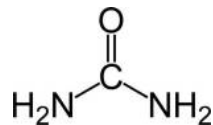
- Polyphenols – Epicatechin, Catechin, etc



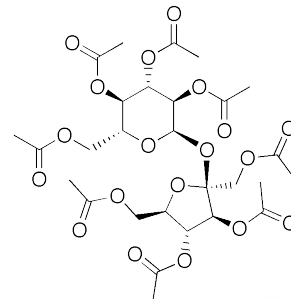
- Quinine



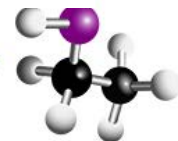
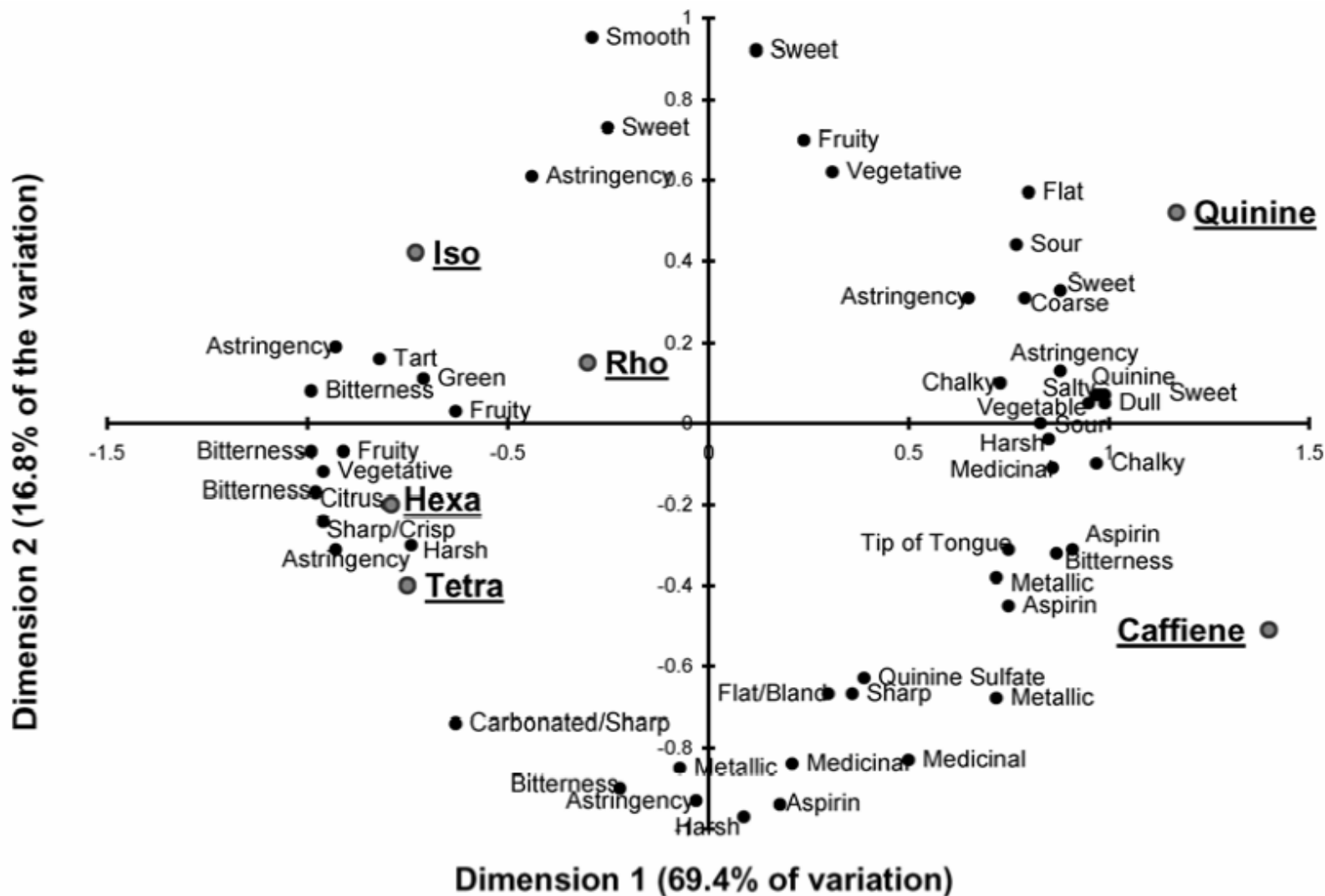
- Urea



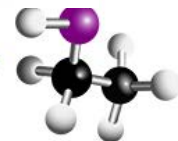
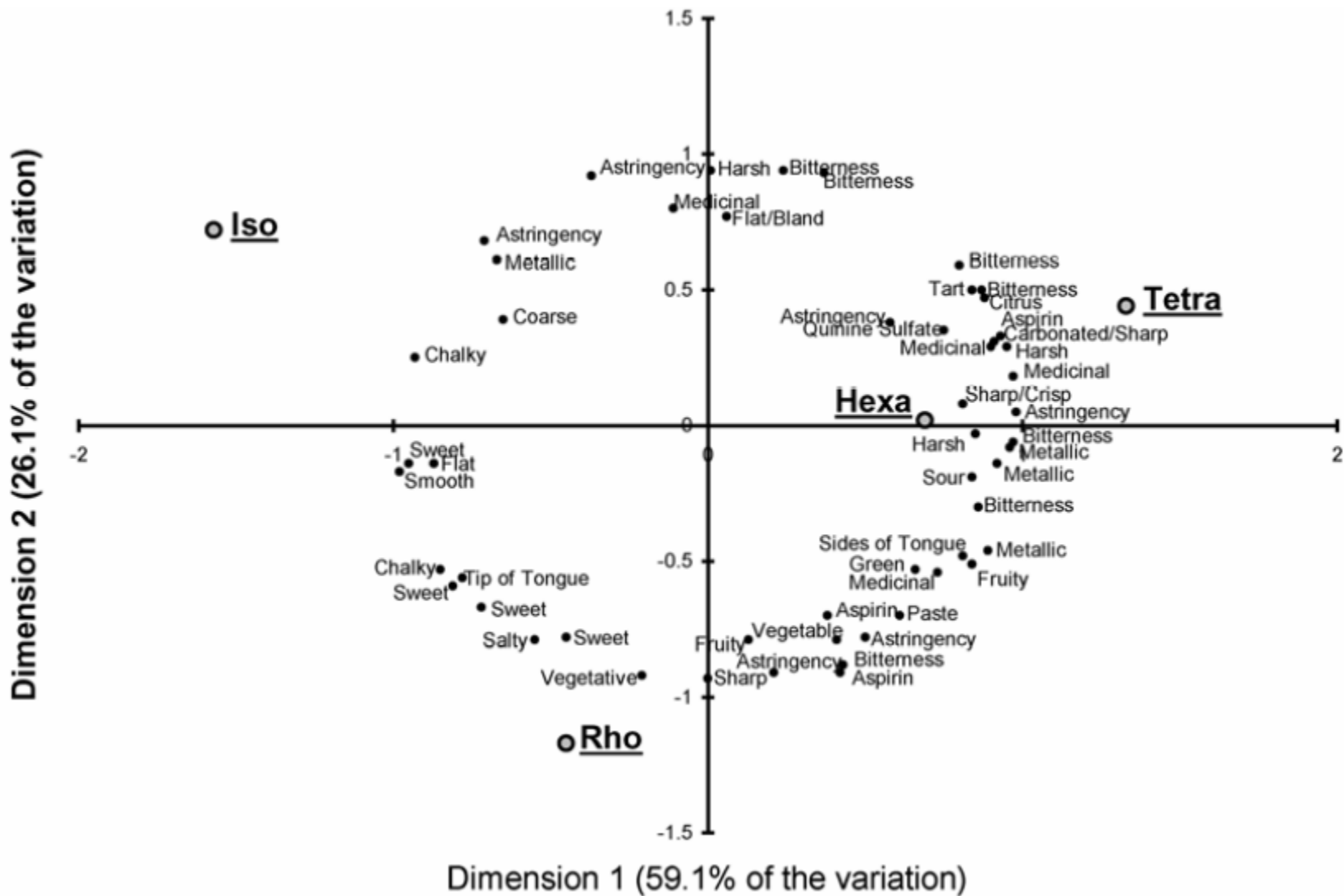
- Sucrose octa-acetate



Bitterness Quality



Bitterness Quality



Bitterness Intensity and Time Intensity

- Applied example



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Modification of perceived beer bitterness intensity, character and temporal profile by hop aroma extract

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<https://doi.org/10.1016/j.foodres.2016.05.018>

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Highlights

- Aroma modified intensity, character and temporal profile of bitterness in beer



Bitterness Perception

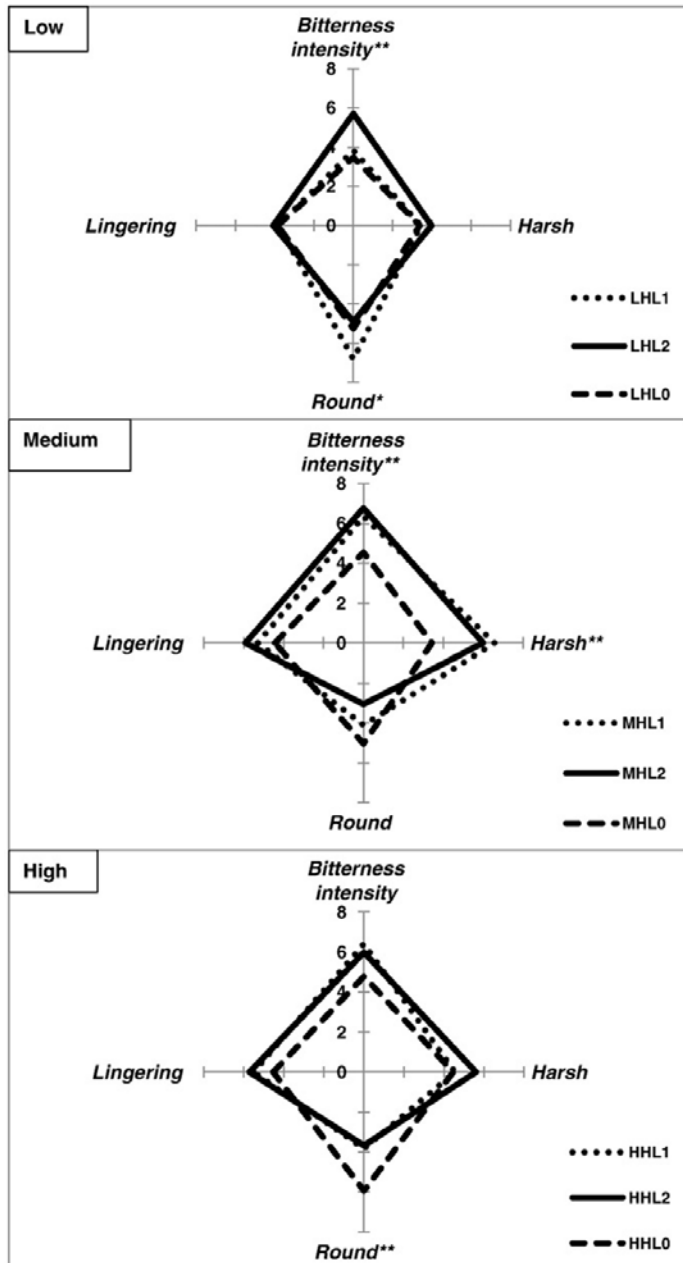
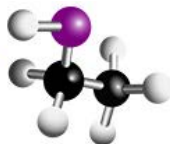


Fig. 1. Spider plots of mean bitterness intensity and bitter character based on intensity ratings. Low: (13 BU) beer, Medium: (25 BU) beer and High: (42 BU) beer. L0, L1 and L2 at each BU level corresponds to hop aroma extract addition levels of 0, 245 and 490 mg/L. Significance denoted at *5% and **1% level.



Time Intensity

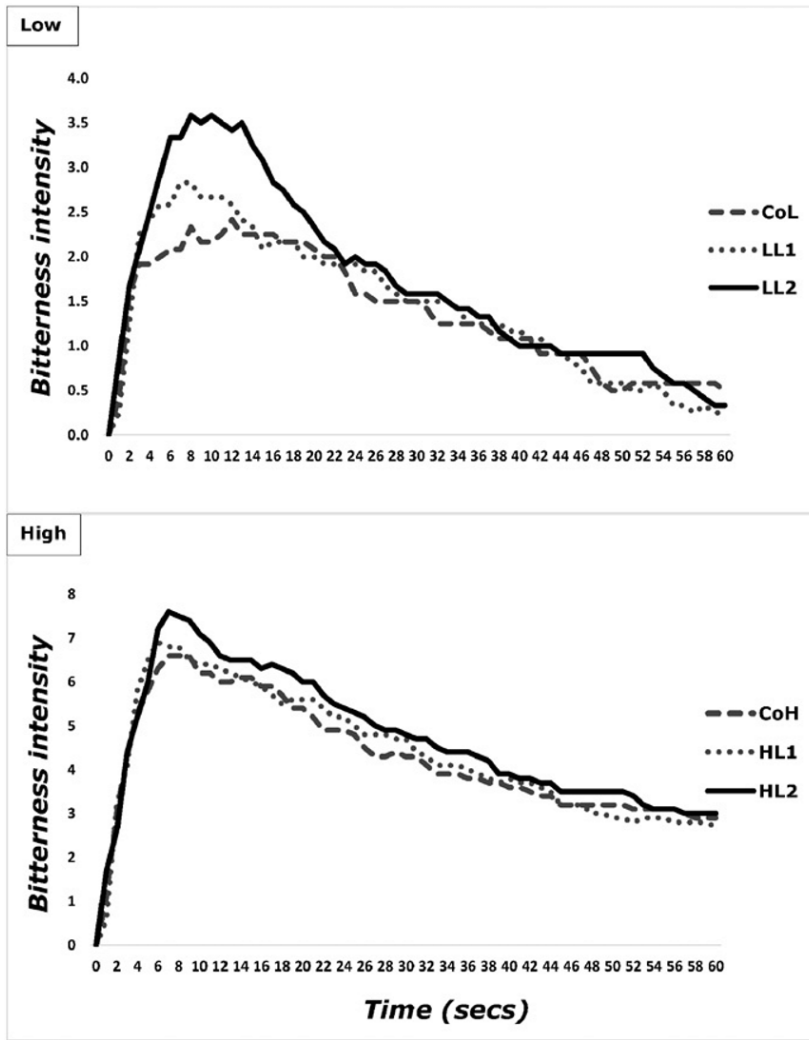
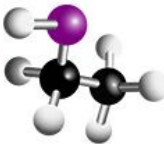


Fig. 4. Average time-intensity curves. Low: (13 BU) beer and High: (42 BU) beer. CoL and CoH, LL1 and HL1, LL2 and HL2 correspond to hop aroma extract addition levels of 0, 245 and 490 mg/L respectively. Significance at 5% level.



Tasting

- Base Beer +10 ppm of Iso
- Base Beer +30 ppm of Iso
- Base Beer +30 ppm of Iso and hop aroma

